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A diagonal relationship is said to exist between certain pairs of diagonally adjacent elements in the second and third periods (first 20 elements) of the periodic table.These pairs (lithium (Li) and magnesium (Mg), beryllium (Be) and aluminium (Al), boron (B) and silicon (Si), etc.) exhibit similar properties; for example, boron and silicon are both semiconductors, forming halides that are ...

Lists of metalloids - Wikipedia

Rayner-Canham G & Overton T 2006, Descriptive Inorganic Chemistry, 4th ed., WH Freeman, New York, (لام در 0-7167-8963-9) Rayner-Canham G 2011, 'Isodiagonality in the Periodic Table', Foundations of chemistry, vol. 13, no. 2, pp. 121–9, doi : 10.1007/s10698-011-9108-y

Jaksollinen järjestelmä – Wikipedia

Występowanie i właściwości. Jest nieznany w stanie czystym i określany bywa także jako wodny roztwór dwutlenku siarki, jednak w roztworze takim występują liczne indywidua chemiczne, głównie różna hydraty SO 2 · nH 2 O, a także jony H 3 O +, HSO − 3, S 2 O 2− 5 i śladowe ilości SO 2− 3.W roztworze nie wykryto natomiast cząsteczek H 2 SO 3 i przyjmuje się, że jeśli w ...

Kwas azotowy – Wikipedia, wolna encyklopedia

Essential Chemistry (2nd ed.). Boston: McGraw-Hill ^ □□□□□-□-□-□□□□□□□ Rayner-Canham G. & Overton T. (2013). Descriptive Inorganic Chemistry (6th ed.). New York: W. H. Freeman and Company • Brown et al. (2009). Chemistry: The Central Science (11th ed.). Upper Saddle River, New Jersey: Pearson Education • Moore et ...

Alt forn - Viquipèdia, l'enciclopèdia lliure

Det periodiske system er en tabelarrangering af grundstofferne, ordnet efter deres atomnumre (antal protoner), elektronkonfigurationer og gennemgående kemiske egenskaber.Denne ordning viser periodiske tendenser, såsom grundstoffer med lignende opførsel i samme søjle.Det viser også fire rektangulære blokke med omtrentligt ens kemiske egenskaber. . Generelt gælder det indenfor enhver ...

Helium – Wikipedia

Tabel periodik adalah tampilan unsur-unsur kimia dalam bentuk tabel. Unsur-unsur tersebut disusun berdasarkan nomor atom (jumlah proton dalam inti atom), konfigurasi elektron, dan keberulangan sifat kimia.Tabel juga terbagi menjadi empat blok: blok -s, -p, -d, dan -f.Secara umum, dalam satu periode (baris), di sebelah kiri bersifat logam, dan di sebelah kanan bersifat non-logam.

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Bảng tuần hoàn (tên đầy đủ là Bảng tuần hoàn các nguyên tố hóa học, còn được biết với tên Bảng tuần hoàn Mendeleev), là một phương pháp liệt kê các nguyên tố hóa học thành dạng bảng, dựa trên số hiệu nguyên tử (số proton trong hạt nhân), cấu hình electron và các tính chất hóa học tuần hoàn của chúng.

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